

Name: \_\_\_\_\_ Date: \_\_\_\_\_ Period: \_\_\_\_\_

PRACTICE - Moles and Grams

DIRECTIONS: This practice test DOES NOT count for any grade. However, take it just as if it were the REAL TEST.

1. 5.25 moles Lead (IV) Acetate = \_\_\_\_\_ g Lead Acetate
2. 2.35 moles Ammonium Sulfate = \_\_\_\_\_ g Ammonium Sulfate
3. 1.45 moles Magnesium Chloride = \_\_\_\_\_ g Magnesium Chloride
4. 9.94 moles Copper (II) Phosphate = \_\_\_\_\_ g Copper (II) Phosphate
5. 15.9 moles Strontium Iodate = \_\_\_\_\_ g Strontium Iodate
6. 88.0 moles Chromium (III) Carbonate = \_\_\_\_\_ g Chromium (III) Carbonate
  
7. 284 g Barium Sulfate = \_\_\_\_\_ moles Barium Sulfate
8. 15.3 g Tin (IV) Permanganate = \_\_\_\_\_ moles Tin (IV) Permanganate
9. 458 g Zinc Chloride = \_\_\_\_\_ moles Zinc Chloride
10. 888 g Beryllium Hypochlorite = \_\_\_\_\_ moles Beryllium Hypochlorite
11. 23.4 g Silver Nitrate = \_\_\_\_\_ moles Silver Nitrate
12. 84.8 g Antimony (V) Carbonate = \_\_\_\_\_ moles Antimony (IV) Carbonate